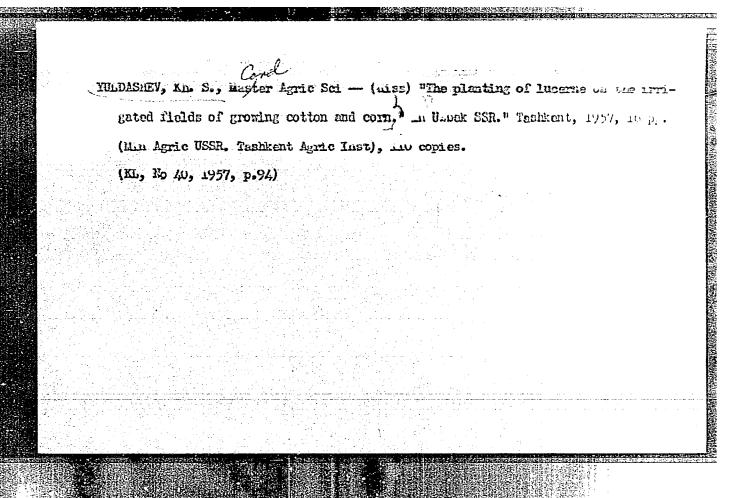
YULDASHEY, Khalil; LANDA, L.W., kend.istor.neuk, otv.red.; AKBAROV, A., red.; SALAKHUTDINOYA, A., tekhred.

[From the history of the development of socialist industry in Tashkent] Iz istorii razvitiia sotsislisticheskoi promyshlennosti Tashkents. Otvetstronnyi red. L.W.,Lenda. Tashkent, Gos. izd-vo Uzbekskoi SSR, 1960. 175 p.

(MIRA 14:1)

(Tashkent--Industries)



YULDASHEV, K.Yu.; TSUKEWANIK, I.P.

Reactions of phenylacetylene and 2-methyl-1-phenylacetylene with anisole. Zhur. ob. khim. 34 no.8:2647-2652 Ag '64. (MIRA 17:9)

1. Tashkentskiy gosudarstvennyy universitet im. V.I. Lenina.

ATAMHANOV, E.I.; KHARAT'YAN, A.H.; BUDYANSKIY, M.V.; YULDASHEV, U.I.; SHAMSUTDINOVA, R.K.; YULDASHEV, K.Yu.

State of some metabolic indices in peptic ulcer of the storach and duodenum and the effect on them of hydrolypate therapy.

Terap.arkh. no.7:85-91 J1 162. (MIRA 15:8)

1. Iz kafedry propedevtiki vnutrennikh bolezney (zav. - chlenkorrespondent AMN SSSR i AN Uzbekskoy SSR prof. E.I. Atakhanov) pediatricheskogo i sanitarno-gigiyenicheskogo fakulitetov Tarikentskogo meditsinskogo instituta. (PEPTIC ULCER) (PROTEIN HYDROLYSATES) (NITROGEN METABOLISM)

# KHARAT'YAN, A.M.; YULDASHEV, K.Yu.

Characteristics of the amino acid composition of protein hydrolysate produced by Central Institute of the Order of Lenin of Hemntology and Blood Transfusion. Probl. gemat. 1 perel, krovi 10 no.2: 52-55 F 164. (MIRA 19:1)

1. Kafedra propedevtiki vmutrennikh bolezney (zav. - chlen-korrespondent AMN SSSR i AN UzSSR prof. E.I. Atakhanov) sanitarn6-gigiyenicheskogo i pediatricheskogo fakul'tetov Tashkentskogo meditsinskogo instituta.

YULDASHEV, K. YU

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Card 3 1 1 30, 152 - 10, 11

Authors : Toukerwonlb, L. F., and Yuliashev, L.

Title : Promination of L,1-lishemylethane

Periodical : Thur. ob. khip. 24 4, 1566-1568, e

Abstract : Bromination of L,1-daymenylethane . . . .

was carried out in a quartz flack ... bulb at 100 - 000°. Heating of the count the ethane hadrocarbons and possequent ... mide. The brownistion products were constallization. The effect of the cobrosination roducts, is explained.

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Clart traces on Central As accelente University

AUTHORS: Zakutskaya, H. A., Yuldashev, Kh. 1 SOY/79-29-2-15/71

TITLE: Condensation of o-Nitroanisole With Chloral Hydrate (Kondensatsiya o-nitroanizola s khloral'gidratom)

PERIODICAL: Zhurnal obshchey khimii, 1959, Vol 29, Nr 2, pp 429-431 (USSR)

ABSTRACT: The condensations of o- and p-nitrophenol (Refs 8,9) and o-nitro-

anisole with chloral hydrate (Ref 10) are known among the direct syntheses of nitrodiaryl trichloroethanes. D. A. Shirley obtained only a yield of 12% 1,1,1-trichloro-2,2-di-(4-methoxy-3-nitro-phenyl)-ethane (II) from o-nitroanisole (0.5 mol) with chloral hydrate (0.25 mol) in the presence of concentrated sulfuric

acid (50 gr) and 20% oleum (100 gr). No secondary products forming in this condensation are mentioned in this connection. On synthesizing trichlorinated carbinols and on investigating their reactions the authors studied this condensation more closely. On

varying the quantity of sulfuric acid and its rate of addition they obtained from nitroanisole (0.2 mol), chloral hydrate (0.1 mol) and sulfuric acid (1.5 mol) the compound (II) in a yield of 60% and a small amount (5%) 1,1,1-trichloro-2,2-(4-methoxy-3-nitrophenyl)(2'-methoxy-3'-nitrophenyl)-ethane (III)

Card 1/2 With a smaller amount of H<sub>2</sub>SO<sub>A</sub> (0.41 mol to 0.1 mol o-nitro-

Condensation of o-Nitroanisole With Chloral Hydrate

SOV/79-29-2-15/71

anisole) the authors succeeded inseparating 6% of the intermediate product (I) (4-methoxy-3-nitrophenyl trichloromethyl carbinol). In some condensations of o-nitroanisole (0.2 mol) with chloral hydrate (0.1 mol) and sulfuric acid (1.5 mol) only traces formed of (I), while compound (IV), melting only at 350°, formed as the chief product. Its exidation product above the positive reaction on an anthraquinone nucleus. With repair to the papers by Quelet (Ref 12) and collaborators the authors suppose the compound (IV) to be the product of autocondensation of (I) and to be a bis-(mesotrichloromethyl)-dimethoxy dinibro dihydro anthracene (Scheme). There are 17 references, 3 of which are Soviet.

ASSOCIATION:

Sredneaziatskiy gosudarstvennyy universitet ((Soviet) Central

Asian State University)

SUBMITTED:

January 8, 1958

Card 2/2

Cond with	ensation of 1-bromo bengene. Uzb. khi	-2-methyl-1-propens a m. zhur. no.6:58-62	nd 1-chloro-1-butene 60. (MIRA 14:1)
1. T 2. 0	ashkentskly gosudar hlen-korrespondent (Propene)	stvennyy universitet AN UzSSR (for TSukerv (Butene)	im. V.I. Lenina. anik). (Benzene)
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1. Tashkentskiy gosudarstvennyy universitet. (Vinyl compounds) (Toluene) (Anisole)

Reactions of chlorostyrene with benzene. Zmur.ob.khim. 32 no.4:
1293-1296 Ap '62. (MIRA 15:4)

1. Tashkentskiy gosudarstvennyy universitet.
(Styrene) (Benzene)

AMIHOV, Alim Muminovich, doktor ekonom.nauk; YULDASHEV, M.Yn., doktor istoricheskikh nauk, red.; AKSKL'ROD, N.B., red.; BAKHTIYAROV, M., tekhred.

[Economic development of Central Asia; from the second half of 19th century to the First World War] Ekonomicheskoe razvitie Srednei Azii; so vtoroi poloviny XIX stoletiia do pervoi mirovoi voiny. Tashkent, Gos.izd-vo UzSSR, 1959. 295 p.

(MIRA 12:8)

(Soviet Central Asia -- Economic conditions)

1. Institut geologii i razrabotki neftyanykh i gazovykh mestorozhdeniy All Uzbekskoy SSR. (Ust-Urt-Bitumen-Geology)	Ritumen potential of Mesozoic sediments in the Ust-Urt. Uzb.geolizhur. 6 no.2:38-45 162. (MIRA 15:4)
	montonophdonia AN Hzbekskov Sil.

# YULDASHEY, P.A.; YUNUSOV, S.Yu, Structure of vincanine. Uzb.khim.zhur. 7 no.1:44-49 '63. (MIRA 16:4) 1. Institut rastitel'nykh veshchestv AN UZSSR. (Vincanine)

MAKSUDO	Chemical investigation of Artemisia scoparia. 6 no.5:84-86 '62.	Uzb.khim.zhur. (MIRA 15:12)					
1. Institut khimii rastitel'nykh veshchesty AN UzSSR. (Uzbekistan-Artemisia)							

YULDASHEV, P.Kn.

YULDASHEV. P. "Investigation of the Alkaloids of Vinca erecta."
Published by the Acad Sci Uzbek SSR. Acad Sci Uzbek
SSR. Inst of Chemistry. Tashkent. 1955.
(DISSERTATION FOR THE DEGREE OF CANDIDATE IN
CHEMICAL SCIENCE).

So.: Knizhnaya letopis' No. 27, July 2, 1955.

TUNUSOV, S.Yu., akademik; YULDANHEV, P.; PIEKHANOVA, N.V.

Study on alkaloids from the absveground portion of Vinca erecta

Rgt. et Schmalh. Dokl. AN Uz. SSR no.7:13-15 '56.

(NIRA 12:6)

1.Akademiya nauk UzSSR (fer Yunusov).

(Alkaleids) (Vinca)

**经验证的运行的证据的证据,就是被除出的行行的证明,并由此对法院的现在,但是实现是对法院的对任任任务的**是,但是实现在是不是,不

In the article, "Reserpicio From Visca erecta," 3. Ya. famasov, Academician of the Academy of Sciences Uzbek SSR and P. Kh. Yuldashev of the Institute of Chemistry, Academy of Sciences Uzbek SSR, describe the method of isolation of the alkaloid reserpinis from Visca erecta, a plant of the Apocynaceae family closely related to the plant Rauvolfia. A total of 2.5 percent of alkaloids are extracted with ether from the roots of the plant. These include the alkaloids vinkania--3.0%2.0%2, vinkadinia--2.0%2.0%3.0%2, and reserpinia--2.2%2.6%1.0%2; reserpinin is saponified with an alkali to form reserpinic acid. The acid and its nitrate are then methylated with diasomethane to obtain the pure alkaloid. (Doklady Akademii Nauk Uzbek-skoy SSR, No 9, 1956, pp 23-25).

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UBAYEV, Kh.; YULDASHEV, P.Kh.; YUNUSOV, S.Yu.

Study of alkaloids of Pedicularis olgae RGL. Uzb.khim.zhur. 7 no.3: 33-36 163. (MIRA 16:9)

Minstitut khimii rastitel'nykh veshchestv AN UzSSR. (Figwort) (Alkaloids)

YAGUDAYEV, M.R.; RASHKES, Ya.V.; YULDASHEV, P.Kh.

Infrared spectra of vincanine and its derivatives. Uzb. khim. zhur. 7 no.6:54-58 '63. (MIRA 17:2)

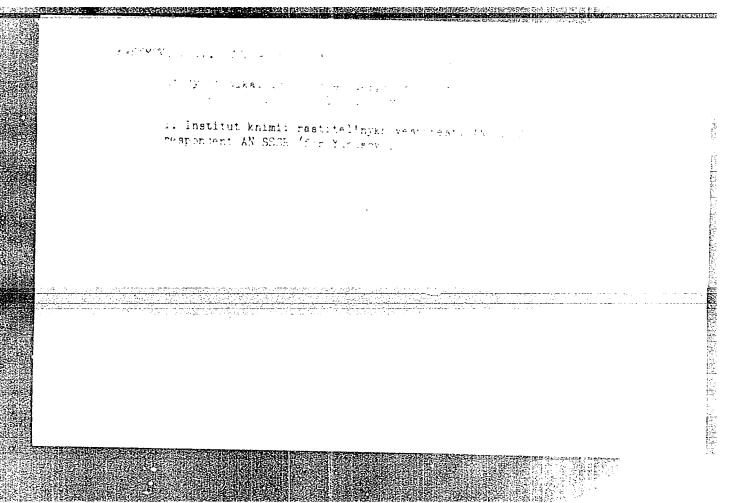
1. Institut khimii rastitel'nykh veshchestv AN UzSSR.

YULDASHEY, P.Kh.; YUNUSOV, S.Yu.

Vincarine, a new alkaloid from the roots of Vinca erecta RGL. et Schrait.

Dokl. AN SSSR 154 no.6:1412-1413 F'64. (MIRA 17:2)

1. Institut khimii rastitel'nykh veshchestv AN UzSSR. 2. Chlen-korrespondent AN SSSR (for Yunusov).

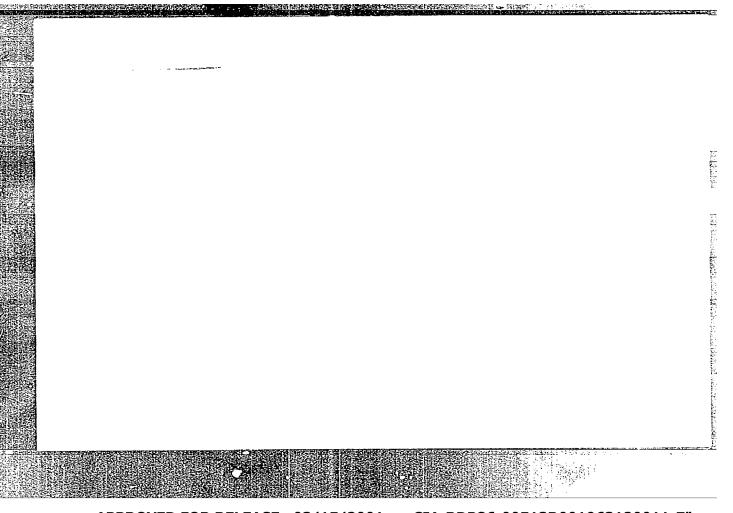


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KASYMOV, Sh.Z.; YULDASHEY, I.Kh.; YUMBSOV, S.Yu.

Structure of vinerine and vineridine. Doll. AN SSSR 1/3 no./:1466. hg 165.

1. Institut khimii rastitel'nykh veshchestv AN UzSSR. 2. Chlen-korrespondent AN SESR (for Yunusov).

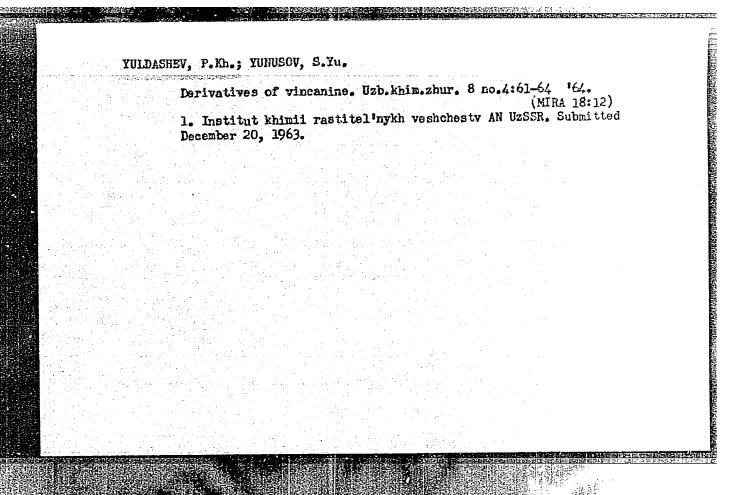


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KUCHRNKOVA, M.A.; YUIDASHEV, P.Kh.; YUNUSOV, S.Yu.

Vinervine, a new alkaloid from the above-ground part of Vinca erecta RGL et Schmalh. Izv.AN SSSR.Ser.khim. no.12:12152-2155 '65. (MIRA 18:12)

1. Institut khimii ras'itel'nykh veshchestv AN U2SSR. Submitted July 29, 1943.



UEAYEV, Kh.; YUDASHEV, P. Kh.; YUMUSOV, S. Yu., akademik

Studying the root alkale is of Vinca erecta Egl. et Schmalh.

Bokl. AN Uz.SSR 21 no. 10:34-37 \*64 (MIRK 19:1)

1. Institut khimii rastitel nykh veshchestw AN Uzzisk. 2. AN Uzsisk (for Yunusov). Submitted May 22, 1964.

Schm 165.	malh roots. Izv.	idine, alkaloi AN SSSR. Ser.	d of vinca e khim. no.ll	recta ngi et :1992-1995 (MIRA 18:11	.)	
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A THOR: Yulin, M. K.; Vol'epshteyn, A. P.
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             The authors describe a processing of alkyl shere?
alkylation of phenol with isobutyl alcohole, developed it om
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and was milicone oil. P-tert-Butylphenol (PTPF) was included
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cethoxyl groups, but melting at a higher to just
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that vinerine and vineridine are diastericleimers
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מס-עונלנ עו INT(n)\_ ACC NA AP6026897 SOURCE CODE: UR/0062/65/000/012/2152/2155 ASTHUR: Kuchenkova, M. A.; Yuldashev, P. Kh.; Yunusov, S. Yu. ORG: Institute of the Chemistry of Vegetable Matter, AN UZSSR (Institut Krimii rastitel'nykh veshchestv AN UzSSR) TITIE: Vinervine -- a new alkaloid from the above ground portion of Vinca erecta SOURCE: AN SSSR. Izvestiya. Seriya khimicheskaya, no. 12, 1965, 2152-2155 TOPIC TAGS: plant chemistry, alkaloid, phenol, molecular structure, melting point ABSTRACT: Specimens of the plant Vinca erecta, collected in Southern Uzbeldstan, were used for cold other extraction of alkaloids, which were then divided into phenolic and nonphenolic fractions. This resulted in the subsequent isolation of a new phenol base, vinervine (C20H22O3N2) (m.p. 154-155°C) which is unstable to light or when in solution, and contains one OCH3 group and two active H atoms one of which is a phenol. This new base caused a pronounced levorotation of polarized light, which points to the presence of a chromophore of alphamethyleneindoline connected to a carbomethoxyl group. Of the three oxygens present in vinervine, one is a phenol hydroxyl and two, are esters. Heating vinervine in ansealed and evacuated ampoule with 15% HC1 for 2.5 hr at 100°C led to the formation of a crystalline indolenine base with a melting point of 185-187°C, which proved to be identical to the indolenine base obtained from vincanidine (another base present in V. erecta). The investigation of the structural position of the phenol hydroxyl is continuing. Orig. art. has: l figure. [JPRS: 36,455] SUB CODE: 07 / SUBM DATE: 29Jul63 / ORIG REF: 002 / OTH REF: 002 Cord 1/1

ACC NR. AP7011362

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SOURCE CODE: UR/0393/66/000/604/0293/0294

AUTHOR: Aripov, Kn. N.; Shakirov, T. T.; Yuldashev, P. Kh.

ORG: Institute of Chemistry of Vegetable Matter, Academy of Sciences USSR (Institut khimii rastitel nykh reshchestv AN UZSSR)

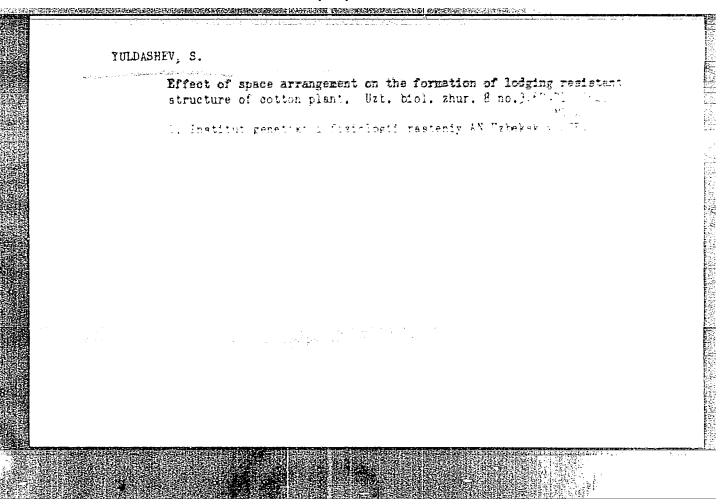
TITLE: Isolation of vincanine

SOURCE: Khimiya prirodnykh soyedineniy, no. 4, 1966, 293-294

TOPIC TAGS: plant chemistry, acetic acid

SUB CODE: 07,06

ABSTRACT: Vincanine was extracted by a countercurrent mothod from the roots of Vinca erecta Rgl. et Schmalh with a 1 percent solution of acetic acid. The extract was desorbed with 1.5 percent ammoniacal solution in 85 percent ethyl alcohol. The evaporated alcoholic solution was acidified with concentrated hydrochloric acid and percent caustic soda, and extracted three times with chloroform. The latter was distilled under vacuum to dryness and, after treatment with acetone, vincanine was isolated and converted into vincanine hydrochloride. [JPRS: 40,351]



	Case of acute psychosis caused by Taeniarhynchus infestation.  Med.zhur.Uxb. no.10:84-85 0 58. (MIRA 13:6)
	1. Is Bukharskoy oblastnoy bol'nitsy (glavnyy vrach - I.I. Aminov).  (TAPEWORKS) (MENTAL ILLINESS)
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YULDASHEV, S.Kh.; AKCHURINA, N.A.

Role of carbohydrates in the lodging of cotton plants. Uzb. blol. zhur. 7 no.6167-73 '63. (MIRA 17:6)

1. Institut genetiki i fiziologii rasteniy AN UzSSR.

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TUIDABLEW, U, (g. Fergana).

The rank of progressive workers is increasing. Pron. keep. 12 no.3:
6 Mr '58.

(MIRA 11:3)

1. Predsedatel' pravleniya oblpromsoveta,
(Fergana Province--Cooperative societies)
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### YULDASHEV, U.I.

Vitamin B-12 level of the blood serum and serum iron in anemias of gastrointestinal origin. Probl. genat. i perel. krovi 5 no.2:2-1? F '60. (MIRA 14:5)

1. Iz kafedry propedevtiki vnutrennikh bolezney (zav. - prof. E.I. Atakhanov) pediatricheskogo i sanitarno-gigiyenicheskogo fakul'tetov Tashkentskogo meditsinskogo instituta.

(CYANGCORALAMINE) (IRON IN THE BODY)

(ANEMIA)

## Concentration of vitamin B<sub>12</sub> and iron in blood serum in diseases of the liver. Med. zhur. Uzb. no.3:10-12 Mr '60. (Mid. 15:2) 1. Iz kafedry propedevtiki vnutrennikh bolezney (zav. - prof. E.I. Atakhanov) pediatricheskogo i sanitarnogo i salitatov Tashkentskogo gosudarstvennogo meditsinskogo instituta. (GYANOCOBALAMINE) (SERUM) (IRON IN THE BODY) (LIVER\_DISEASES)

ATAKHANOV, E.I., prof.; YULDASHEV, U.I., assistent

Amount of vitamin B<sub>12</sub> and iron in blood serve in storach diseases. Mod. zhur, Uzb. no.4:8-13 Ap '61. (MIRA 14:5)

l. Iz kafedry propedevtiki vnutrennikh bolezney pediatricheskogo i sanitarno-gigiyenicheskogo fakul'tetov Tashkentskogo gosudarstvennogo meditsinskogo instituta. (STOMACH\_\_DISEASES) (SERUM) (IRON IN THE BODY) (CYANOCOBALAMINE)

ATAKHANOV, E.I.; KHARAT'YAN, A.M.; HUDYANSKIY, M.V.; YULDASHEV, U.I.; SHAHSUTDINOVA, R.K.; YULDASHEV, K.Yu.

State of some metabolic indices in peptic ulcer of the stomach and duodenum and the effect on them of hydrolysate therapy.

Terap.arkh. no.7:85-91 Jl 162. (MIRA 15:8)

1. Iz kafedry propedevtiki vnutrennikh bolezney (zav. - chlenkorrespondent AMN SSSR i AN Uzbekskoy SSR prof. E.I. Atakhanev) pediatricheskogo i sanitarno-gigiyenicheskogo fakul'tetov Tashkentskogo meditsinskogo instituta. (PEPTIC ULCER) (PROTEIN HYDROLYSATES) (NITROGEN METABOLISM)

### YULDASHEV, U. I.

Dynamics of the vitamin B<sub>12</sub> and iron content in the blood sermin the pellagra syndrome. Terap. arkh. 34 no.5:71-76 '62.

(MIRA 15:6)

1. Iz kafedry propedevtiki vnutrennikh bolezney (zav. - chlenkorrespondent AMN SSSR i AN UzSSR prof. E. I. Atakhanov) pediatricheskogo i sanitarnogo fakulitetov Tashkentskogo meditsinskogo instituta.

(PELLAGRA) (CYANOCOBALAMINE) (IRON IN THE BCDY)

WULDASHEY, U.I., kand.med.nauk

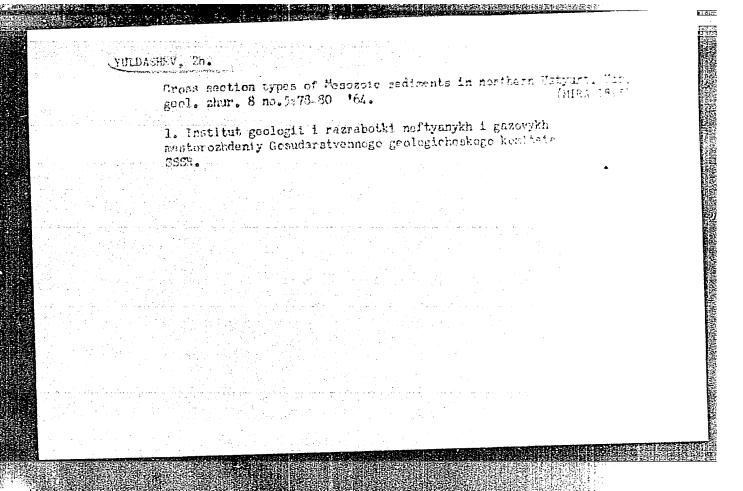
Content of vitamin B<sub>2</sub> and iron in the blood serum in anemia.
Terap. arkh. 34 ho.12:69-74 D'62. (MIRA 16:6)

1. Iz kafedry propedevtiki vnutrennikh bolezney (zav. - chlen.
'orrespondent ANN SSER i AN UZSER prof. E.I.Atakhanov) podiatricheskogo i sanitarno-gigiyenicheskogo fakul'tetov Tashkentskogo
meditsinskogo instituta.
(CYANOCOBALAMINE) (IRON IN THE BODY)

### YULDASHEV, Zh.

Characteristics of the petrographic composition of Mesozcic sediments in the northern Ustyurt. Uzb.geol.zhur. 8 no.3:34-38 '64. (MTRA 18:12)

1. Institut geologii i razrabotki neftyanykh i gazovykh mestorozhdeniy Gosudarstvennogo geologicheskogo komiteta SSSR. Submitted Jan. 31, 1964.



ACCESSION NR: AT4042432

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S/3103/64/000/002/0175/0182

AUTHOR: Usmanov, Kh. U., Tillayev, R. S., Musayev, U. N., Yuldasheva, Kh.

TITLE: Thermomechanical properties and plasticizing of grafted copolymers obtained by radiation polymerization

SOURCE: AN UZSSR. Institut khimii polimerov. Khimiya i fiziko-khimiya prirodny\*kh i sinteticheskikh polimerov, no. 2, 1964, 175-182

TOPIC TAGS: grafted copolymer, acrylonitrile, polystyrene, polyvinylchloride, vinyl perchloride, glass temperature, Gamma-irradiation, plasticizer, saponified copolymer, radiation polymerization, polymer plasticizing, polymer thermomechanical property

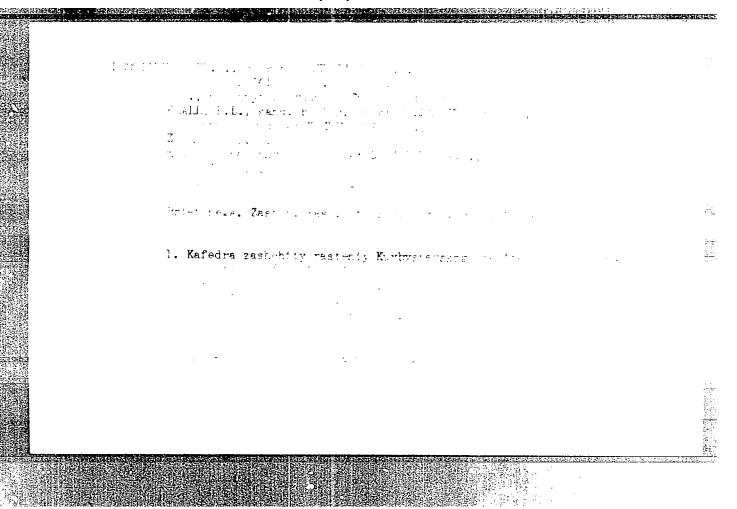
ABSTRACT: A study of the thermomechanical properties of grafted copolymers obtained by grafting acrylonitrile on polystyrene, polyvinvyl chloride and vinyl perchloride showed that the glass temperature To of these copolymers, regardless of the ratio of the components, corresponds essentially to the glass temperature of the initial polymers, but that the flow temperature To lies above the temperature of chemical stability of the products. Copolymers, as compressed tablets (3-4 mm thick and 7 mm in diameter), were tested before and after irradiation at doses of 1-10 Mr. The thermomechanical curves were plotted with the dynamometric scales of Kargin and Sogolova at a constant load for 10 sec., at a specific

ACCESSION NR: AT4042432

load of 1.4 kg/cm<sup>2</sup>. The curves obtained for all the copolymers, with or without plasticizers were quite similar, and showed less effect of temperature than on pure polymers. Tabulated irradiation data showed that the thermomechanical properties of grafted copolymers remain almost unchanged under the influence of irradiation. This indicates the greater stability of grafted copolymers to Y-rays as well as to high temperatures. The flow of grafted copolymers is therefore considered to be almost independent of grafting. An investigation of the plasticizing of grafted copolymers showed that grafted copolymers synthesized from two homopolymers which have a common plasticizer remain unchanged in their compatibility with this plasticizer. For grafted copolymers containing, on the one hand, chains able to plasticize (polystyrene, polyvinyl chloride) and, in the other component, unplasticizable rigid chains (polyacrylonitrile), the compatibility with the plasticizer is low and limited. The change in thermomechanical properties (decrease in Tc) with increasing plasticizer concentration (tetralin or methylbenzoic ether) is plotted. In addition, analytical data for nitrogen content and acid number of the grafted copolymers are tabulated. The thermomechanical curves of saponified vinyl perchloride and polyacrylonitrile grafted copolymers showed that the glass temperature is decreased and the plasticity is increased by saponification. A further increase in plasticity is produced by plasticizers. especially glycerol. Such an increase could never be obtained by plasticizing unsaponified grafted copolymers. Orig. art. has: 2 tables and 3 figures.

2/3

ACCESSION NR: AT4042432			:
ASSOCIATION: Institut khimi AN UzSSR)	li polimerov AN UzSSR (Institute	oi Polymer Chemistry,	
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YULPASHEVA, L.K.

Category: USSR / Physical Chemistry - Electrochemistry

B-12

Abs Jour: Referat Zhur-Khimiya, No 9, 1957, 30129

Author : Fayzullin F. F., Yuldasheva L. K.

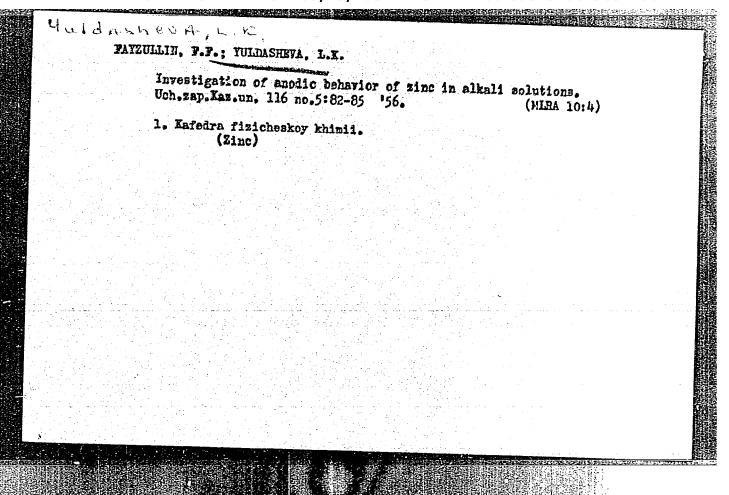
: Kazan' University

Title : Study of Ancdic Behavior of Zinc in Alkaline Solutions

Orig Pub: Uch. zap. Kazanskogo un-ta, 1956, 116, No 5, 82-85

Abstract: By the method of automatic recording of the (  $\omega$  , t) curves (RZhKhim, 1957, 12280) a study was made of anodic polarization of Zn in 0.25, 0.5 and 1 N NaOH at 40 and  $60^{\circ}$  and i=6 g/dm<sup>2</sup>. Cn application of the current the portential of Zn rises sharply and evolution of O2 begins. Cxidation is attended by periodical, very rapid, potential changes, caused by periodical breakdown and formation of oxide film. By the gravimetric method a determination was made of the rate of formation of oxide film on Zn at 1 of 6 and 12 a/dm2; an increase of i increases rate of formation of the film. On increase of the temperature there takes place a decrease in overvoltage of 0, evolution, which results in an increased rate of formation of the oxidic film.

Card : 1/1



8/048/63/027/001/030/043 B125/B102

AUTHORS:

Arbuzov, B. A., Samitov, Yu. Yu., and Yuldasheva, L. K.

TITLE:

Spectra of proton magnetic resonance of the substituted dislocations

PERIODICAL:

Akademiya nauk SSSR. Izvestiya. Seriya fizicheskaya, v. 27,

no. 1, 1963, 89 - 92

TEXT: A study of the p.m.r. spectra of 2-methyldioxolane, 2-chloromethyldioxolane, and trichloromethyldioxolane proved that the influence of the halide replacing the hydrogen in the methyl radical of 2-methyl-1,3-dioxolane extends as far as the protons of the methylene groups that are in 6-position with respect to the oxygen. The polar groups also cause chemical shifts of the \$6-hydrogens. Owing to the effect of the five-membered rings the chemical shifts of the protons in dioxalane are by 0.3 smaller than in the compounds with open chains (e.g. actal, orthoester). A substitution of the proton of the methyl radical by the first chlorine atom influences the chemical shift of the prodons of the methylene group more strongly than the subsequent introduction of further chlorine atoms. Card 1/2

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	Spectra	of proton	y (	S/048/6 B125/B10	<b>3/027/0</b> 01/030/043 92	
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	Card 2/2	<u>-4, # (</u>			era	

Reaction of selenols with acetylene derivatives. Part 2:
Structure of the products of interaction between selections and phenylacetylere and semethylenetyle

# ARBUZOV, A.; YULDASHEVA, L.K. Dipole moments and the conformation of cyclic compounds. Report Real: 1,3-Diexolanes. Izv. AN SSSR.Otd.khim.nauk no.10:1728-1734 | 1... (Mind. 1)... 1. Khimichekiy institut im. A.M.Butlerova Kazanskogo gosudarstveno pouniversiteta. (Dioxolane-Dipole moments)

### ARBUZOV, B.A.; YULDASHEVA, L.K.

Dipole moments and the conformation of cyclic compounds. Izv. AN SSSE, Otd.khim.nauk no.10:1734-1737 0 '62. (MIRA 15:10)

1. Khimicheskiy institut im. A.M.Butlerova Kazanskogo gosudarstvennogo universiteta.

(Diexans-Dipole momenta)

ARBUZOV, B. A.; SAMITOV, Yu. Yu.; YULDASHEVA, L. K.

Proton magnetic resonance spectra of substituted dioxolenes. Izv. AN SSSR, Ser. fiz. 27 no.1:89-92 Ja '63.

(MIRA 16:1)

1. Maranskiy gosudarstvennyy universitet im. V. I. Uliyanova-Lenina.

(Nuclear magnetic resonance and relaxation)
(Dioxolanes—Spectra)

ARBUZOV, B.A., ukademik; VILICHINSKAYA, A.R.; SAMITOV, Yu.Yu.; YULDASHEVA, L.K.

Structure of allocalmene dioxide. Dokl. AN SSSR 164 no.5:1041(MIRA 18:10)

1. Nauchno-issledovatel skiy khimicheskiv institut im. A.M. Butlerova pri Kazanskom gosudarstvennom universitete.

KHABIBULLIN, Sh.T.; YULDASHEVA, L.L.				
	Analysis of star coun Ogorodnikov's method.	ts in the dark neculae Uch.zap.Kaz.un. 116	using K.E. no.1:89-92 (MIRA 10	:5)
	1. Kafedra astronomii.	(Neblulae) (Ogorodni)	cov, K.E.)	

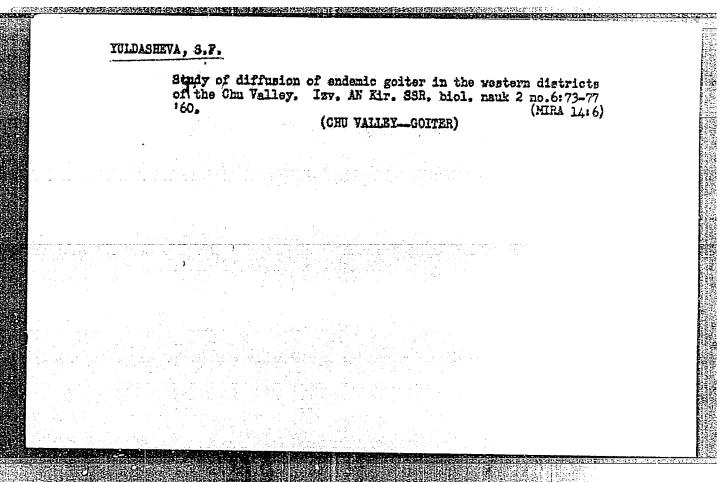
### YULDASHEVA, M.

Selecting the design form of the systems of equations for unit voltages in electric systems. Izv. AN Uz. SSR. Ser. tekh. nauk 9 no. 5:23-25 '65. (MIPA 18:10)

1. Nzbekskiy nauchno-issledovatel skiy institut energetiki i avtomatiki.

YULDASHEVA, O. Kak moye zveno dobilos' urozhaya sakharnoy svekly po 831 tsentinery s gektara. (kog'khoz kzyl kuyash. kok-andsk, rayon fergan. obl. sots. sel. khoz-vo uzbekictana, 1949, No 3, s. 56-60

SO: Letopis' Zhurnal' nykh Statey, Vol. 44



	services for the	unatic children	in Tashkent.	Pediatriia (NLRA 9:8)
nauki prof. tuta imeni (RH	ki gospitalinoy R.S.Gershenovic V.M.Holotova EDMATISH, in in management in sa	ch) Tashkentskog Sant and child,	, zasluzhennyy go meditsinsko	deyatel' go insti-

## TUIDASHEVA, S.H. Distribution and clinical forms of rheumatic fever in children in Uzbekistan. Pediatrils 35 no.12:18-20 D '57. (MIRA 11:2) 1. Is kliniki gospital'noy pediatril Teshkentskogo meditsinskogo instituta (zev. - zesluzhennyy deyatel' nauki prof. R.S.Gershenovich) (UZBEKISTAN--RHEUMATIC FEVER)

LYUBETSKAYA, M.Z.; YULDASHEVA, S.N.; NURLDDINOV, M.R.

Conditioned reflex changes in the pupil in rheumatic fever in children. Pediatriia 36 no.2:89 F 159. (MIRA 12:4)

1. Iz kliniki gospital'noy pediatrii Tashkentakogo meditsinskego instituta.

(PUPIL (EYE)) (RHEUMATIC FEVER)

BUSSEL', L.G.; YULDASHEVA, S.N.

Hemocultures of streptococci in cases of rheumatic fever in children. Pediatrina 39 no.1:55-60 '61.

MIFA

l. Iz kafedry mikrobiologii (zav. - prof. P.F. Samsonov), kafedry bolezney ukha, gorla i nosa (zav. - prof. N.Yu. Laskov) i gornov pediatricheskoy kliniki (zav. - prof. R.S. Geraden v.c.: Tashkentskogo meditsinskogo instituta.

(RHEUMATIC FEVER) (STREPTOCOGCUS)

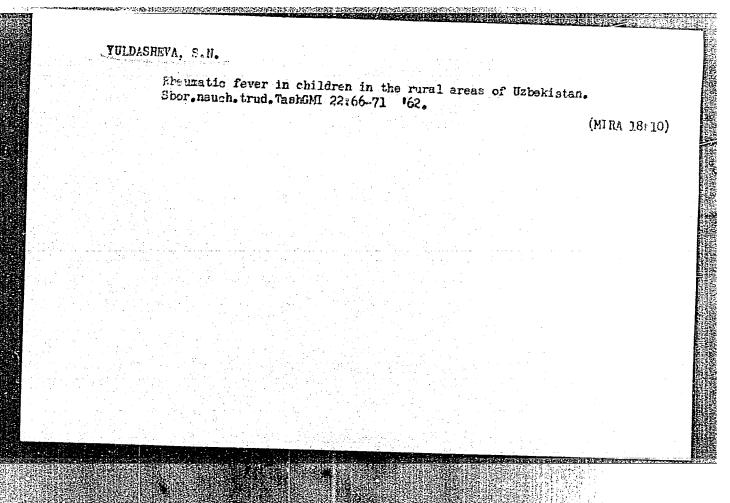
YULDASHEVA, S.N., dotsent

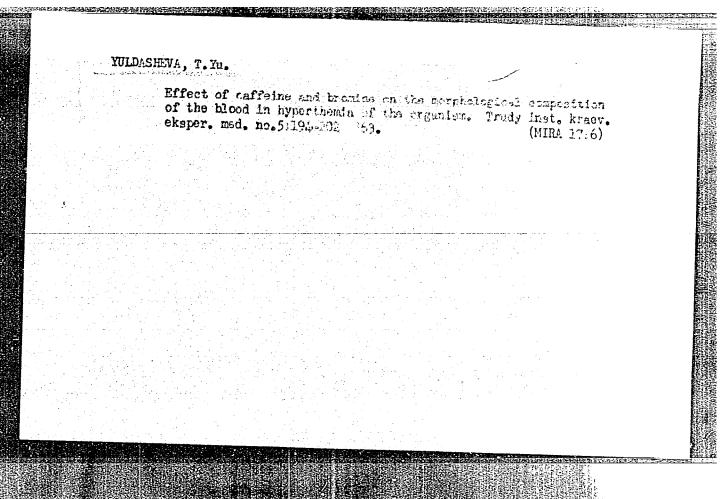
Decrease of rheumatic fever in children in same cities and rural localities of Uzbekistan. Trudy Tadzh. med. inat. 50:34-56 '661.

(Mid: 17:8)

1. Zaveduyushchaya kafedroy faxul'tetskoy pediatrii Tushkantskop meditsinskogo instituta.

# Distribution of rheumatism among children in rural areas of Uzbekistan. Vop. okh. mat. i det. 7 no.5:57-59 My '62. (MIRA 15:6) 1. Iz gospital'noy pediatricheskoy kliniki Tashkentskogo meditsinskogo instituta (zav. - prof. R.S. Gershenovich [deceased]). (UZEEKISTAN-RHEUMATIC FEVER)





기는 이 환경 수석 보는 것으로 받았다. 	Effect of overheating and insolation on the blood picture and reticulocyte content in the blood of healthy and decorticated dogs. Uzb. blol. zhur. 7 no.5:53-55 '63.				
	(MIRA 18:11)  1. Tashkentskiy gosudarstvennyy universitet imeni Lenina.				

S/078/61/006/004/003/018 B121/B216

AUTHORS:

Sokolova, N. D., Skuratov, S. M., Shemonayeva, A. M.

Yuldasheva, V. M.

TITLE:

Determination of the standard enthalpy of formation of the

alpha and beta modification of metaboric acid

PERIODICAL:

Zhurnal neorganicheskoy khimii, v. 6, no. 4, 1961, 774-776

TEXT: The standard enthalpies of formation of the alpha and beta modifications of metaboric acid were obtained by determining the standard enthalpies of solution at  $295^{\circ}$ K.  $\alpha$ -HBO $_2$  was prepared by heating analytical grade  $H_3BO_3$  for several days in an ampulla under a vacuum of 10-20 mm Hg at  $90^{\circ}$ C.  $\beta$ -HBO $_2$  was obtained by heating boric acid in an open ampulla to  $160^{\circ}$ C in the course of 8 hr and keeping it at this temperature for several days. X-Ray analytical data indicated the products to be the pure  $\alpha$ - and  $\beta$  modifications. X-Ray analysis was made by A. A. Babad-Zakhryapin at the Institut fizicheskoy khimii Akademii nauk SSSR (Institute of Physical

Card 1/3

Determination of the standard ...

S/078/61/006/004/003/018 B121/B216

Chemistry, Academy of Sciences USSR). The measurements were carried out in a calorimeter with an adiabatic jacket. Metaboric acid was introduced into the calorimeter in closed ampullas which were then broken. The thermometer readings were correct to  $\pm$  0.0005°. The water equivalent of the calorimeter was determined by electrical heating ( $\sim$  171 cal/deg). The temperature rise was 0.03-0.06°C for  $\alpha$ -HBO<sub>2</sub>, and 0.17°C for  $\beta$ -HBO<sub>2</sub>. The enthalpy of solution of  $\alpha$ -HBO<sub>2</sub> was measured to be 700 and 400 mole H<sub>2</sub>O for a final concentration of 1 mole H<sub>3</sub>BO<sub>3</sub>, both values agreeing within the measuring error. For  $\beta$ -HBO<sub>2</sub>, the enthalpy of solution was measured at a final concentration of 1 mole H<sub>3</sub>BO<sub>3</sub> to 500 mole H<sub>2</sub>O. The enthalpies of formation of the alpha and beta modifications of metaboric acid determined at final concentrations of 1 mole H<sub>3</sub>BO<sub>3</sub> to 500 mole H<sub>2</sub>O

are  $\alpha - \text{HBO}_2$   $\Delta H_{293} = + 0.47 \pm 0.01 \text{ kcal/mole}$  $\beta - \text{HBO}_2$   $\Delta H_{293} = + 1.76 \pm 0.01 \text{ kcal/mole}$ 

The standard enthalpies of formation of the alpha and beta modifications

Card 2/3

Determination of the standard ...

S/078/61/006/004/003/018 B121/B216

of metaboric acid from crystalline boron and gaseous oxygen and hydrogen were calculated at  $\alpha$ -HBO<sub>2</sub>  $\Delta H_{\text{formation}}^{O} = -189.0 \pm 0.4 \text{ kcal/mole}$ 

 $\beta$ -HBO<sub>2</sub>  $\triangle$ H<sup>O</sup>formation = -190.3  $\pm$  0.4 kcal/mole

There are 2 tables and 10 references: 3 Soviet-bloc and 7 non-Soviet-bloc.

ASSOCIATION:

Moskovskiy gosudarstvennyy universitet im. M. V. Lomonosova,

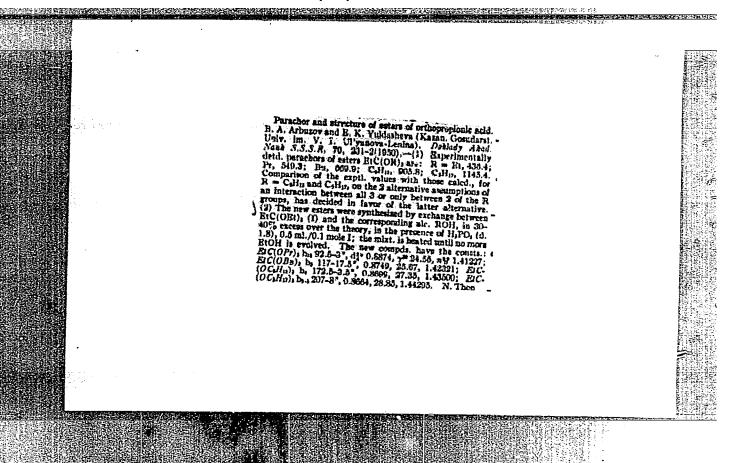
Khimicheskiy fakultet (Moscow State University imeni

M. V. Lomonosov, Chemical Division)

SUBMITTED:

March 4, 1960

Card 3/3



Use of Academician H.P. Tushnov's histolysates in veterinary practice. Veterinaria 31 no.2:39-40 F '54. (MLRA 7:2)				
1. Bashkirskaya veterinarnaya opytnaya stantsiya. (Veterinary medicine) (Tissue extracts)				
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YULENBER, G. YE.

AUTHOR:

YULENBEK, G. Ye.

53-3-5/10

TITLE: PERIODICAL: "In Memory of Professor P. KHRMIFEST", Russian.

Uspekhi Fiz. Nauk, 1957, Vol 62, Nr 3, pp 367-370 (U.S.S.R.)

ABSTRACT:

On the occasion of being awarded the OERSTED medal, G.E. WILEMEK delivered a speech before the American Union of Teachers of Physics in which he spoke about the great pedagogue and physicist EMRENFEST whose pupil he had been and to whom he owes his

pedagogical successes.

ASSOCIATION:

Not given

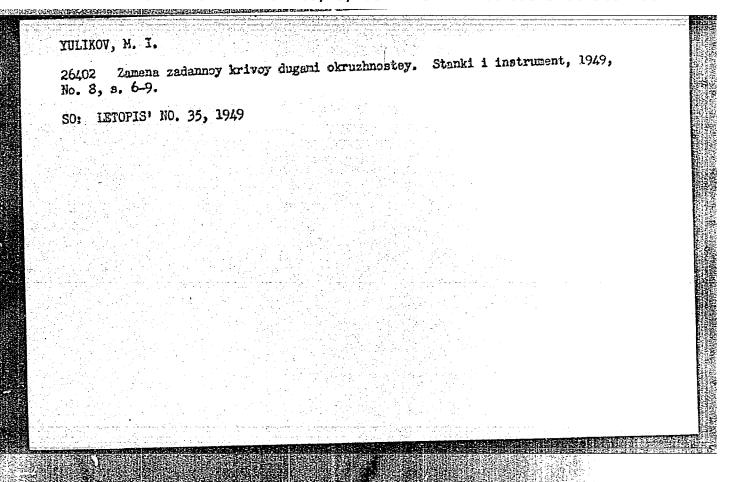
PRESENTED BY:

SUBMITTED:

AVAILABLE:

Library of Congress

Card 1/1



SO: Letopi	is! Zhurnal! nykh	Statey, Vol.	, щ, Moskva,	1949	
				jaran Santajaran	

"Investigation of Rams for Machining Gurvilinear Profiles." Thesis for degree of Cand. Technical Sci. Sub 4 Jun 50, Moscow Order of Labor Red Banner Higher Technical School imeni N. E. Bauman

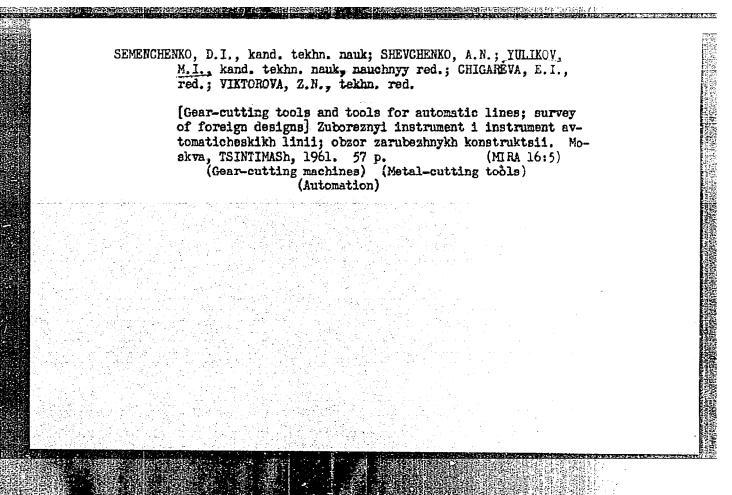
FOR Summary 71, 4 Sep 52, <u>Dissertations Presented</u>
for <u>Degrees in Science ond Engineering in Moscow</u>
in 1950. From <u>Vechernyaya Moskya</u>, Jan-Dec 1950

TEMCHIN, Grigoriy Il'ich; LUR'YB, G.B., prof., retsenzent; YULIKOV, M.I., knd.tekhn.nauk, red.; MCRCZOVA, M.B., red.izdatel'stva; MATYEYEVA, Ye.B., tekhn.red.; EL'XIND, V.D., tekhn.red.

[Theory and computation for setting up multiple-tool equipment]
Teorila i raschet mnogoinstrumentnykh naladak. Moskva, Gos.
nauchno-tekhn.izd-vo mashinostroit.lit-ry, 1957. 555 p.

(Machine tools)

(Machine tools)



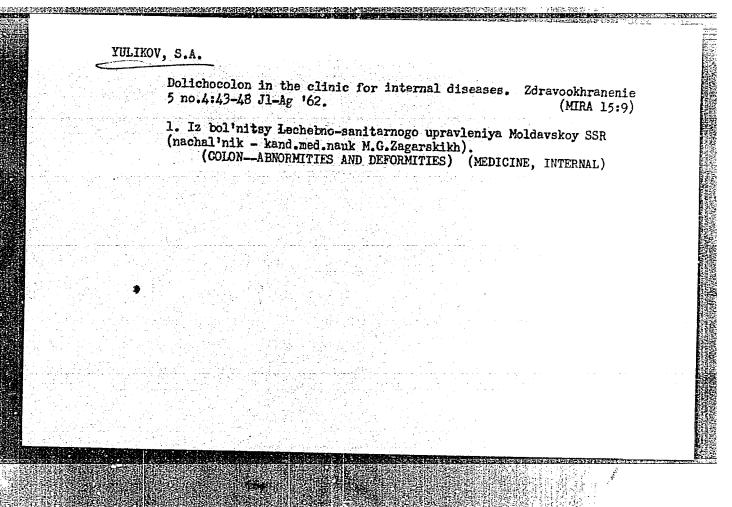
TEMCHIN, G.I.[doceased]; YULIKOV, M.I., kand. tekhn. nauk, red.; retsenzent; ESTENZON, M.A., kand. tekhn. nauk, red.; SEMENCHENKO, V.A., red.izd-va; MODEL', B.I., tekhn. red.; DEMKINA, N.F., tekhn. red.

[Multitool adjustments; theory and design] Mnogoinstrumentnye naladki; teorila i raschet. Izd.2., ispr. strumentnye naladki; teorila i raschet. Izd.2., ispr. (MIRA 16:12) (Moskva, Mashgiz, 1963. 542 p. (Metal cutting)

ARSHINOV, V.A., kand. tekhn. nauk; AIEKSEYEV, G.A., inzh.; YEGOROV, S.V., kand. tekhn. nauk, dots., retsenzent; MALIMOVSKIY, V.R., inzh., retsenzent; TULIKOV, M.I., kand. tekhn.nauk, red.

[Metal cutting and metal-cutting tools] Rezanie metallov i rezhushchii instrument. Koskva, Izd-vo "Mashinostroenie," 1964. 543 p.

(MIRA 17:7)



的重要的更加的企业。但是他们的专门的证明的目光,但是他们的是一种的证明,这个可能是是是他们的是一种,但是他们是一种的。

KHOXHLOV, A.S.; SILAYEV, A.B.; STEPANOV, V.M.; YULIKOVA, Yo.P.; TROSHKO, Ye.V.; LEVIN, Ye.D.; MAMIOPE, S.M.; SINITSYNA, Z.T.; CHI CHAN-TSIN [Ch'ih Ch'ang-Ch'ing]; SOLOV'YEVA, N.K.; IL'INSKAYA, S.A.; ROSSOVSKAYA, V.S.; DMITRIYEVA, V.S.; SEMENOV, S.M.; VEYS, R.A.; BEREZINA, Ye.K.; RUBTSOVA, L.K.

A new type of polymyxin, polymyxin M. Antibiotiki 5 no.1:3-9 Ja-F '60. (MIRA 13:7)

1. Vsesoyuznyy nauchno-issledovatel skiy institut antibiotikov i laboratoriya khimii belka i antibiotikov khimicheskogo fakul'teta Moskov kogo ordena Lenina gosudarstvennogo universiteta imeni M.V. Lomonosova.

(POLYMIXIN)

YULIKOVA, YE. P., KUZMIHA, N. A., SILAYEV, A. B., KATRUKHA, G. S. (USSR)				
"Mechanism of Polymixin M Inactivation."				
Report presented at the 5th International Biochemistry Congress, Moscow, 10-16 August 1961				
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하는 경험 전문 기업				

SILAYEV, A.B.; STEPANOV, V.M.; YULIKOVA, Ye.P.; TROSHKO, Ye.V.; LEVIN, Ye.D.

Chemistry of polymyxin M. Part 1: Qualitative amino acid analysis and analysis for end groups. Zhur. ob. khim. 31 no.1:29-305 Ja 161.

1. Moskovskiy gosudarstvennyy universitet.

(Polymyxin)

SHAYEV, A.B.; STEPANOW, V.M.; YULIKOVA, Ye.P.; MPRATOVA, G.L.

Chemistry of polymixin M. Part 2: Quantitative enine acid composition. Zhur. ob. khim. 31 no.3:1023-1026 Hr '61. (MIRA 14:3)

1. Moskovskiy gosudarstvennyy universitat.

(Polymixin)

Chemistry of polymyxin H. Part 3: Partial hydrolymis of polymyxin H. Zhur.ob.khim. 31 no.8:2712-2716 Ag '61. (MIRA 14:8)				
l. Moskovskiy Lomonosova.	gosudarstvenny (Polymyxin)	y universitet imeni	Ŀ M.V.	
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SILAYEV, A.B.; STEPANOV, V.M.; YULIKOVA, Ye.P.; MICHAYLOVA, I.Yu.; (Bolgariya); UDALOVA, T.P.

Study of the inactivation of polymyxin. M. Antibiotiki 7 no.7: 638-643 J1:62. (MIRA 16:10)

1. Laboratoriya khimii belka i antibiotikov khimicheskogo fakul'teta Moskovskogo universiteta imeni M.V. Lomonosova.

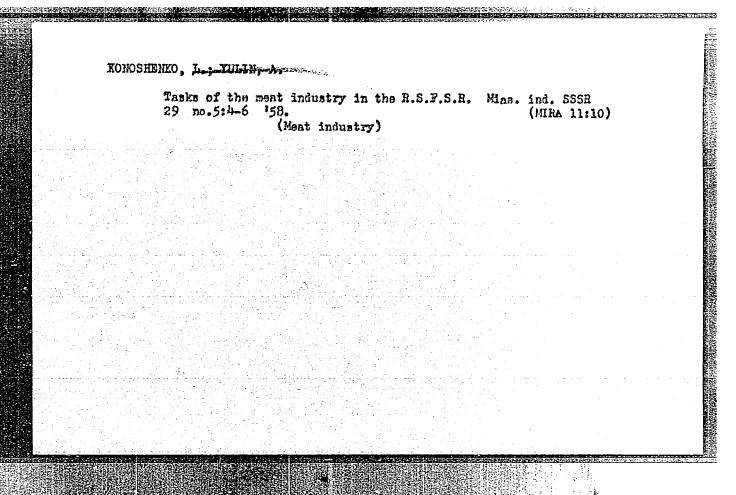
SILAYEV, A.B.; YULIKOVA, Ye.P.; BARATOVA, L.A.

Chemistry of polymyxin M. Part 5: Identification of fatty acid.

Zhur.ob.khim. 32 no.3:818-820 Mr '62. (MIRA 15:3)

THE PERSON OF TH

1. Moskovskiy gosudarstvennyy universitet imeni M.V.Lomonosova. (Polymyxins) (Acids, Fatty)



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VOL'-EPSHTKYN, A. B.; GRIGOR'YEV, S. M.; KRICHKO, A. A.; KOMYASHINA, R. A.; SUROVISEVA, V. V.; YULIN, M. K.

Production of aromatic hydrocarbons from pyrolysis tar of hydrocarbon gases by hydrogenation, Trudy IGI 17:269-277 '62.

(MIRA 15:10)

(Hydrocarbons) (Coal-tar products)
(Hydrogenation)

YULIN M.K.; VOL'-EPSHTEYN, A.B.; DAVTYAN, N.A.; LISYUTKINA, L.N.

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AUTHOR: Makarova, T. F.; Moshkov, P. F.; Sheehin, M. Yulin, M. K.

TIPLE: A method for the preparation of p-tert-butyling

SOURCE: Pyulleten' izobreteniy i tovarnykh znakov.

TOPIC TAGS: text butylphenol synthesis, sulforsted

ABSTRACT: The preparation of p-tert-butylphene investand tri-tert-butylphenols, in the presence of ac action and income violar of the usin production.